

1-[(1*R*,2*R*)-2-(1,8-Naphthalimido)cyclohexyl]-
3-phenylthioureaXue-Mei Yang^a and En-Qin Fu^{b*}^aDepartment of Chemistry, Guangdong Medical College, Dongguan 523808, People's Republic of China, and ^bDepartment of Chemistry, Wuhan University, Wuhan 430072, People's Republic of ChinaCorrespondence e-mail:
xuemeiyang131@163.com

Key indicators

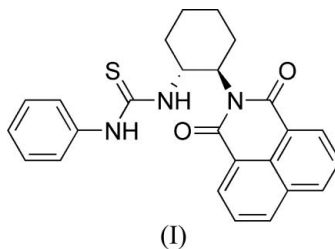
Single-crystal X-ray study
T = 173 K
Mean $\sigma(\text{C}-\text{C}) = 0.004 \text{ \AA}$
R factor = 0.044
wR factor = 0.102
Data-to-parameter ratio = 16.7For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title compound, $\text{C}_{25}\text{H}_{23}\text{N}_3\text{O}_2\text{S}$, the asymmetric unit consists of two molecules connected by $\text{N}-\text{H}\cdots\text{S}$ hydrogen bonds. The dihedral angles between the phenyl and naphthalimide ring systems in the scorpion-like molecules are $60.63(7)$ and $61.09(8)^\circ$. The cyclohexane rings adopt chair conformations.

Received 13 March 2007
Accepted 29 March 2007

Comment

Chiral recognition of carboxylic acids is an interesting and challenging topic as they are important structural components of many natural products and drugs (Coppola & Schuster, 1997; Hanessian *et al.*, 1983; Maier *et al.*, 2001; Valentine *et al.*, 1980). In a search for novel chiral CSAs (chiral solvating agents) (Chinchilla *et al.*, 1995; Parker & Taylor, 1987) for the recognition of chiral carboxylic acids, we have designed and synthesized the title compound, (I), and its structure is reported here.



The asymmetric unit of (I) (Fig. 1) consists of two molecules connected by $\text{N}3-\text{H}3\text{A}\cdots\text{S}2$ and $\text{N}6-\text{H}6\cdots\text{S}1$ intermolecular hydrogen bonds (Table 1). There is also an $\text{N}-\text{H}\cdots\text{O}$ intramolecular hydrogen bond between the cyclohexylamine H atom and a carbonyl O atom of the naphthalimide group in both molecules. The molecules adopt a scorpion-like shape with dihedral angles $60.63(7)$ and $61.09(8)^\circ$ between the phenyl and naphthalimide ring planes of the two molecules. The cyclohexane rings exhibit chair conformations.

In the crystal structure, molecules link into pairs through $\text{N}-\text{H}\cdots\text{S}$ hydrogen bonds. Adjacent pairs interlock, forming chains running along the *a* axis (Fig. 2). These chains pack into columns down the *a* axis (Fig. 3).

Experimental

Phenyl isothiocyanate (0.59 g, 0.0044 mol) in dichloromethane (10 ml) was added slowly to (1*R*,2*R*)-1-(1,8-naphthalimido)-2-aminocyclohexane (1.30 g, 0.0044 mol) in dichloromethane (40 ml) and the mixture stirred at room temperature for 5 h. The solvent was

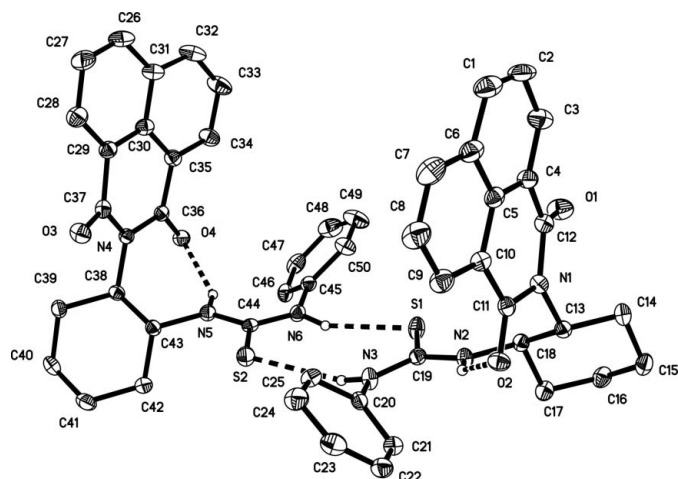


Figure 1

The asymmetric unit of (I), with displacement ellipsoids drawn at the 30% probability level, showing the hydrogen-bonding interactions. Hydrogen bonds are represented as dashed lines. H atoms not involved in hydrogen bonds have been omitted for clarity.

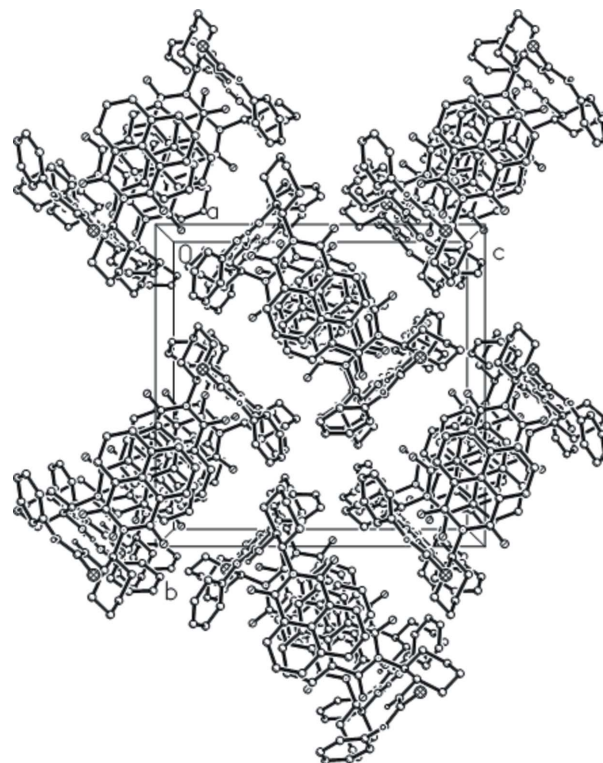


Figure 3

The crystal packing of (I). H atoms have been omitted for clarity.

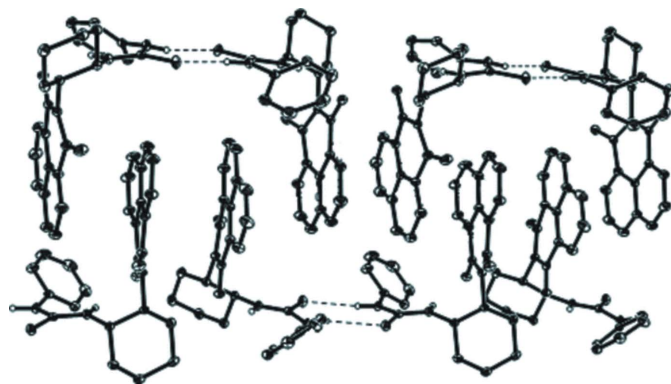


Figure 2

One-dimensional chains along the *a* axis. Hydrogen bonds are represented as dashed lines. H atoms not involved in hydrogen bonds have been omitted for clarity.

evaporated to dryness under reduced pressure and the remaining residue recrystallized from ethanol to afford 1.61 g of yellow needle-like crystals (yield 85%). Crystals suitable for X-ray diffraction were obtained by slow evaporation of an ethanol solution.

Crystal data

$C_{25}H_{23}N_3O_2S$ $V = 4325.9 (4) \text{ \AA}^3$
 $M_r = 429.52$ $Z = 8$
 Orthorhombic, $P2_12_12_1$ Mo $K\alpha$ radiation
 $a = 14.8211 (7) \text{ \AA}$ $\mu = 0.18 \text{ mm}^{-1}$
 $b = 16.8568 (8) \text{ \AA}$ $T = 173 (2) \text{ K}$
 $c = 17.3150 (8) \text{ \AA}$ $0.41 \times 0.30 \times 0.27 \text{ mm}$

Data collection

Bruker AXS SMART 1000 CCD diffractometer 22367 measured reflections
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996) 9357 independent reflections
 $T_{\min} = 0.931$, $T_{\max} = 0.954$ 6724 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.039$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.044$ $\Delta\rho_{\text{max}} = 0.36 \text{ e \AA}^{-3}$
 $wR(F^2) = 0.102$ $\Delta\rho_{\text{min}} = -0.21 \text{ e \AA}^{-3}$
 $S = 1.05$ Absolute structure: Flack (1983),
 9357 reflections 4114 Friedel pairs
 559 parameters Flack parameter: 0.04 (6)
 H-atom parameters constrained

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N2-H2A\cdots O2$	0.88	2.11	2.876 (3)	145
$N3-H3A\cdots S2$	0.88	2.50	3.306 (2)	152
$N5-H5\cdots O4$	0.88	2.17	2.928 (3)	145
$N6-H6\cdots S1$	0.88	2.46	3.295 (2)	159

All H atoms were positioned geometrically and refined using a riding model with, $C-H = 0.95 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for aromatic, $C-H = 1.00 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for CH, $C-H = 0.99 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for CH_2 , and $N-H = 0.88 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{N})$.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT-Plus (Bruker, 2003); data reduction: SAINT-Plus; program(s) used to solve structure: SHELXTL (Bruker, 2005); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

The authors thank the National Science Foundation of China for financial support.

References

- Bruker (2001). *SMART*. Version 5.054. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (2003). *SAINT-Plus*. Version 6.45. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (2005). *SHELXTL*. Version 6.14. Bruker AXS Inc., Madison, Wisconsin, USA.
- Chinchilla, R., Foubelo, F., Najera, C. & Yus, M. (1995). *Tetrahedron Asymmetry*, **6**, 1877–1880.
- Coppola, G. M. & Schuster, H. F. (1997). *Hydroxyl Acids in Enantioselective Synthesis*. Weinheim: VCH.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- Hanessian, S. (1983). *Total Synthesis of Natural Products: The Chiron Approach*. Oxford: Pergamon.
- Maier, N. M., Franco, P. & Lindner, W. (2001). *J. Chromatogr. A*, **906**, 3–33.
- Parker, D. & Taylor, R. J. (1987). *Tetrahedron*, **43**, 5451–5456.
- Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
- Valentine, D. Jr, Johnson, K. K., Priester, W., Sun, R. C., Toth, K. & Saucy, G. (1980). *J. Org. Chem.* **45**, 3698–3703.